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Noriko Ishida^a; Kimiaki Imafuku^a

^a Department of Chemistry, Faculty of Science, Kumamoto University, Kumamoto, Japan

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TAUTOMERISM OF 2-MERCAPTO-8*H*-CYCLOHEPTA[*d*]THIAZOL-8-ONE

NORIKO ISHIDA and KIMIAKI IMAFUKU*

Department of Chemistry, Faculty of Science, Kumamoto University, Kurokami, Kumamoto 860, Japan

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The UV, ¹H NMR, and ¹³C NMR spectra of 2-mercapto-8*H*-cyclohepta[*d*]thiazol-8-one (2) are extremely near to those of 3-methyl-8-oxo-3,8-dihydrocyclohepta[*d*]thiazoline-2-thione (6). These results proved that the thiazoline form 2a predominates in the tautomeric equilibrium of 2.

Key words: 2-Mercapto-8H-cyclohepta[d]thiazol-8-one, 8-oxo-3,8-dihydrocyclohepta[d]thiazoline-2-thione, tautomerism, UV spectra, ¹H NMR spectra, ¹³C NMR spectra.

INTRODUCTION

It has been proposed¹ that 2-mercapto-8*H*-cyclohepta[d]thiazol-8-one (2) exists in tautomeric thiazoline form 2a. Namely, the UV spectrum of the compound 2 is different from that of 2-methylthio-8*H*-cyclohepta[d]thiazol-8-one (3). However, another isomeric 3-methyl-8-oxo-3,8-dihydrocyclohepta[d]thiazoline-2-thione (6) is unknown up to date.

In this paper, we prepared N-methyl-substituted isomer 6 and proved that the tautomerism of the parent compound 2 lies in the thiazoline form 2a by the UV, ¹H NMR, and ¹³C NMR measurements.

RESULTS AND DISCUSSION

Synthesis of 3-Methyl-8-oxo-3,8-dihydrocyclohepta[d]thiazoline-2-thione (6)

2-Mercapto-8H-cyclohepta[d]thiazol-8-one (2)¹ was prepared by the reaction of 3-aminotropolone (1) with carbon disulfide. The treatment of the compound 2 with methyl iodide gave 2-methylthio-8H-cyclohepta[d]thiazol-8-one (3) but did not give N-methyl isomer 6.¹ The methylation of the compound 2 with dimethyl sulfate or methyl fluorosulfate also afforded the S-methyl derivative 3 as a sole product.

Then, 3-aminotropolone $(1)^{2-4}$ was treated with methyl fluorosulfonate to afford 3-methylaminotropolone $(4)^5$ and 7-methylamino-2-methoxytropone (5). A methanolic solution of the compound 4 and carbon disulfide was refluxed for 3 h to afford 3-methyl-8-oxo-3,8-dihydrocyclohepta[d]thiazoline-2-thione (6) in 83% yield. Its structure was confirmed by IR and ¹H NMR spectra as well as elemental analysis.

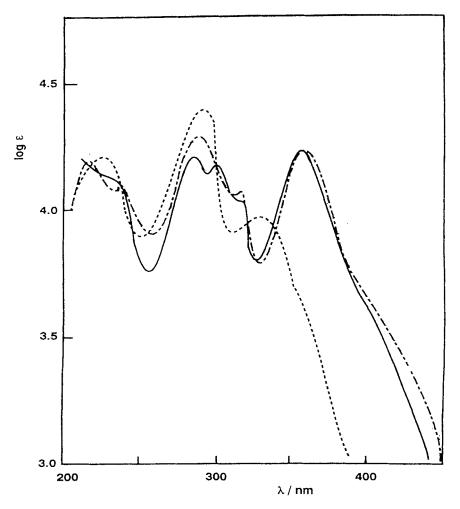


FIGURE 1 The UV spectra of compounds 2 (----), 3 (----), and 6 (- \cdot - \cdot -).

Tautomerism of 2-Mercapto-8H-cyclohepta[d]thiazol-8-one (2)

It was proposed that hydrogen bonding ability and self-association make to shift the tautomeric equilibrium of 2-mercaptothiazoles to thiazoline-2-thiones. In the chemistry of 2-mercapto-8*H*-cyclohepta[*d*]thiazol-8-one (2), it was reported that its UV spectrum is different from that of the 2-methylthio derivative 3, as shown in Figure 1. These results might provide an evidence for tautomeric predominance of the compound 2 over another tautomer 2. However, the fixed *N*-methyl derivative of compound 2 was unknown so far.

Now, we newly prepared N-methyl isomer 6, as described above, and measured its UV spectrum. When the UV spectrum of the N-methyl isomer 6 is drawn in Figure 1, its spectrum is very similar to that of the parent compound 2. Consequently, it is doubtless that 2-mercapto-8H-cyclo-hepta[d]thiazol-8-one (2) exists as 8-oxo-3,8-dihydrocyclohepta[d]thiazoline-2-thione (2a).

The ¹H and ¹³C NMR spectral data of 2-mercapto-, S-methyl-, and N-methyl derivatives, 2, 3, and 6, are shown in Figure 2. The spectral assignments were done by ¹H-¹H and ¹H-¹³C COSY measurements. The ¹H and ¹³C NMR spectral data of the parent compound 2 are nearer to those of the N-methyl isomer 6 than those of the S-methyl isomer 3.

FIGURE 2 The ¹H and ¹³C NMR spectra of compounds 2, 3, and 6.

Particularly, the δ_c values at the 2- and 3a-positions of the compounds 2 and 6 are very similar each other. These 2- and 3a-carbon atoms exist at the both side of the nitrogen atom. Thus, the ¹H and ¹³C NMR spectral data also support the thiazoline form 2a.

In conclusion, it was revealed by the UV, ¹H NMR, and ¹³C NMR spectral data of the fixed S- and N-methyl isomers that 2-mercapto-8H-cyclohepta[d]thiazol-8one (2) exists predominantly as the thiazoline form 2a.

EXPERIMENTAL

The melting points were determined with a Yanagimoto MP-S2 micromelting point apparatus and are uncorrected. The IR spectra were taken on a JASCO A-102 spectrophotometer, the UV spectra on a Hitachi 2000 spectrophotometer. The ¹H and ¹³C NMR spectra were measured on a JEOL JNM-GX400 spectrometer (400 MHz) and partly on a JEOL JNM-PMXSI60 spectrometer (60 MHz). The mass spectra were obtained with a JEOL JMS-DX300 spectrometer.

2-Mercapto- (2) and 2-Methylthio-8H-cyclohepta[d]thiazol-8-one (3)

These compounds were obtained according to the methods described in the literature.

- 2: Mp 295°C (dec) [lit, 1255–300°C (dec)]; IR (KBr): ν_{max} 1616 cm⁻¹ (C=O); UV (CH₃OH): λ_{max} 227 sh (log ε 4.13), 284 (4.20), 297 (4.17), 356 nm (4.22); H NMR (DMSO- d_0): δ 7.02 (1H, d, J = 12.1 Hz, 7-H), 7.26 (1H, dd, J = 11.0, 8.1 Hz, 5-H), 7.32 (1H, dd, J = 11.0, 1.5 Hz, 4-H), 7.40 (1H, dd, J = 12.1, 8.1 Hz, 6-H); ¹³C NMR (DMSO- d_6): δ 123.62 (4-C), 132.91 (5-C), 135.91 (6-C), 137.01 (7-C),140.33 (8a-C), 143.65 (3a-C), 175.71 (8-C), 190.41 (2-C).
- 3: Mp 113–114°C (lit, 112–113°C); IR (CHCl₃): ν_{max} 1628 cm⁻¹ (C=O); UV (CH₃OH): λ_{max} 227 (log ε 4.20), 285 sh (4.36), 291 (4.37), 329 nm (3.94); ¹H NMR (CDCl₃): δ 2.78 (3H, s, CH₃), 7.04 (1H, dd, J = 11.0, 8.8 Hz, 5-H), 7.08 (1H, d, J = 12.5 Hz, 7-H), 7.29 (1H, dd, J = 12.5, 8.8 Hz, 6-12.5 Hz, 7-H), 7.20 (1H, dd, J = 12.5, 8.8 Hz, 6-12.5 Hz, 7-H), 7.20 (1H, dd, J = 12.5 Hz,H), 7.74 (1H, d, J = 11.0 Hz, 4-H); ¹³C NMR (CDCl₃): δ 16.39 (CH₃), 129.57 (5-C), 132.85 (4-C), 136.00 (6-C), 136.23 (7-C), 140.74 (8a-C), 153.83 (3a-C), 174.98 (8-C), 179.08 (2-C); MS: m/z (%) 209 (M*, 100), 181 (32), 166 (5), 148 (55), 135 (13), 108 (27).

Reaction of 3-Aminotropolone (1) with Methyl Fluorosulfonate

To a solution of 3-aminotropolone (1) (247 mg, 1.8 mmol) in dichloromethane (1.0 ml) was added methyl fluorosulfonate (1.0 ml). The mixture was allowed to stand for 12 h at room temperature. The resulting mixture was diluted with water, treated with 1M sodium hydroxide solution (5 ml), and extracted with chloroform. The evaporation residue was chromatographed on a Wakogel B-10 plate $(30 \times 30 \text{ cm})$ with ethyl acetate-hexane (1:1) to give 3-methylaminotropolone (4) and 7-methylamino-2-methoxytropone (5).

- 4: Yield 108 mg (36%); yellow plates (from ethanol); mp 113-114°C (lit, 13.5-115°C).
 5: Yield 7 mg (2%); oil; H NMR (CDCl₃, 60 MHz): δ 3.05 (3H, s, NCH₃), 3.95 (3H, s, OCH₃). 6.4-7.4 (4H, m); MS: m/z (%) 165 (M⁺, 100), 137 (63), 122 (21). Found: M⁺, 165.0801. Calcd for C₉H₁₁NO₂: M, 165.0790.

Preparation of 3-Methyl-8-oxo-3,8-dihydrocyclohepta[d]thiazoline-2-thione (6)

To a solution of 3-methylaminotropolone (4) (108 mg) in methanol (70 ml) containing potassium hydroxide (2.13 g) was added carbon disulfide (3.6 ml). The resulting mixture was refluxed for 3 h. The reaction mixture was concentrated and acidified with 4M hydrochloric acid to give 3-methyl-8-oxo-3,8-dihydrocyclohepta[d]thiazoline-2-thione (6).

6: Yield 125 mg (83%); yellow crystals (from chloroform); mp 260-261°C; IR (KBr): ν_{max} 1627 cm⁻¹ (C=O); UV (CH₃OH): λ_{max} 212 (log ε 4.17), 241 sh (4.04), 285 (4.27), 297 sh (4.19), 312 (4.07), 357 nm (4.21); ¹H NMR (DMSO- d_0): δ 3.89 (3H, s, CH₃), 7.10 (1H, d, J = 12.1 Hz, 7-H), 7.36 (1H, dd, J = 11.7, 8.8 Hz, 5-H), 7.49 (1H, dd, J = 12.1, 8.8 Hz, 6-H), 7.65 (1H, d, J = 11.7 Hz, 4-H); ¹³C NMR (DMSO- d_0): δ 34.91 (CH₃), 124.98 (4-C), 132.57 (5-C), 136.16 (6-C), 136.88 (7-C), 138.86 (8a-C), 144.02 (3a-C), 175.59 (8-C), 189.34 (2-C); MS: m/z (%) 209 (M+, 100), 181 (32), 148 (37), 136 (18). Found: C, 51.31; H, 3.28; N, 6.68%. Calcd for C₉H₇NOS₂: C, 51.65; H, 3.37; N, 6.69%.

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REFERENCES

- 1. S. Seto and K. Ogura, Bull. Chem. Soc. Jpn., 37, 1526 (1964).
- 2. J. W. Cook, J. D. Hobson and D. K. Steel, J. Chem. Soc., 530 (1954).
- 3. T. Nozoe and Y. Kitahara, Proc. Jpn. Acad., 30, 204 (1954).
- 4. K. Imafuku, M. Furuya and Z.-T. Jin, Bull. Chem. Soc. Jpn., 57, 609 (1984).
- 5. K. Kikuchi and Y. Maki, Bull. Chem. Soc., Jpn., 51, 2338 (1978).
- 6. a) E. Gentric, J. Lauransan, C. Roussel and J. Metzer, J. Chem. Soc., Perkin Trans., 2, 565 (1976);
 - b) E. Gentric, J. Lauransan, C. Roussel and J. Metzer, J. Chem. Soc., Perkin Trans., 2, 1016 (1977).